Amphidinolide Y, a Novel 17-Membered Macrolide from Dinoflagellate *Amphidinium* **sp.: Plausible Biogenetic Precursor of Amphidinolide X**

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Abstract: A novel cytotoxic 17-membered macrolide, amphidinolide Y (**1**), has been isolated from a marine dinoflagellate *Amphidinium* sp., and it was elucidated to exist as a 9:1 equilibrium mixture of 6-keto- and 6(9)-hemiacetal forms (**1a** and **1b**, respectively) on the basis of 2D NMR data and chemical means. The feeding experiments with 13Clabeled acetates suggested that amphidinolide Y (**1**) may be a biogenetic precursor of 16-membered macrodiolide, amphidinolide X (**2**).

In our continuing search for bioactive secondary metabolites from laboratory-cultured marine dinoflagellates,¹ a novel cytotoxic 16-membered macrodiolide, amphidinolide X (**2**), has been isolated from a dinoflagellate *Amphidinium* sp. (strain Y-42), which is a symbiont of Okinawan marine acoel flatworm *Amphiscolops* sp.2 Amphidinolide X (**2**) is the first macrodiolide consisting of polyketide-derived diacid and diol units from natural sources. Our search for biosynthetic precursors of this unique macrodiolide resulted in the isolation of a novel 17-membered macrolide, designated amphidinolide Y (**1**), from the same strain. Here, we describe the isolation and structure elucidation of **1** and its labeling patterns with acetates.

The dinoflagellate *Amphidinium* sp. (strain Y-42) was separated from a marine flatworm *Amphiscolops* sp. collected off Sunabe, Okinawa. The dinoflagellate was mass cultured unialgally at 25 °C for 14 days in a seawater medium enriched with 1% ES supplement and $13C$ -labeled NaHCO₃. The harvested algal cells (315 g, wet weight, from 500 L of culture) were extracted with MeOH/toluene (3:1), and the extracts were partitioned between toluene and water. The toluene-soluble materials were subjected to a silica gel column (CHCl₃/MeOH) followed by C_{18} HPLC (CH₃CN/H₂O) to afford amphidinolide Y (**1**, 0.0007%, wet weight) together with known

macrolides, amphidinolides G^3 (0.0008%), H^3 (0.0007%), W4 (0.009%), and X2 (**2**, 0.0004%). In the 13C NMR spectrum, 10% enrichments for all carbon signals of **1** were observed.

Amphidinolide Y (**1**) had the molecular formula of $C_{26}H_{42}O_6$ as revealed by HRESIMS [m/z 473.2875 (M + Na)⁺, -0.4 mmu]. IR absorptions at 3450 and 1711 cm⁻¹ were attributed to hydroxyl(s) and carbonyl group(s), respectively. 1H and 13C NMR data of **1** (Table 1) in CDCl3 disclosed the existence of a ketone, an ester carbonyl, an $sp²$ quaternary carbon, three $sp²$ methines, two oxygenated $sp³$ quaternary carbons, five $sp³$ methines (three of which were oxygenated ones), seven $sp³$ methylenes, and six methyl groups (three of which resonated as a singlet signal each due to connection to a quaternary carbon). In the 1H NMR spectrum of **1**, a set of proton resonances were observed in a ratio of $9:1,5$ while the ¹³C NMR spectrum showed some minor signals including a hemiacetal carbon (δ _C 104.56), indicating that 1 existed as a 9:1 equilibrium mixture of 6-keto and 6(9)-hemiacetal forms (**1a** and **1b**, respectively). Since four out of six unsaturations were accounted for, the 6-keto form (**1a**) of amphidinolide Y (**1**) was inferred to contain two rings. Interpretation of the 1H-1H COSY, TOCSY, and HMQC spectra revealed proton connectivities of the following units: (a) from H-2 to H_2 -5 and H_3 -22, (b) from H_2 -8 to H-11 and H₃-24, (c) from H₂-13 to H₂-17, and (d) from H_2 -19 to H_3 -21 (Figure 1). ¹H and ¹³C NMR data of three partial structures, **a**, **c**, and **d**, in **1a** were similar to those of the corresponding portions of amphidinolide X (**2**).

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⁽¹⁾ Kobayashi, J.; Shimbo, K.; Sato, M.; Tsuda, M. *J. Org. Chem.*

²⁰⁰², *⁶⁷*, 6585-6592 and references therein. (2) Tsuda, M.; Izui, N.; Shimbo, K.; Sato, M.; Fukushi, E.; Kawabata, J.; Katsumata, K.; Horiguchi, T.; Kobayashi, J. *J. Org. Chem.* **2003**, *⁶⁸*, 5339-5345.

^{(3) (}a) Kobayashi, J.; Shigemori, H.; Ishibashi, M.; Yamasu, T.; Hirota, H.; Sasaki, T. *J. Org. Chem.* **¹⁹⁹¹**, *⁵⁶*, 5221-5224. (b) Kobayashi, J.; Shimbo, K.; Sato, M.; Shiro, M.; Tsuda, M. *Org. Lett.* **²⁰⁰⁰** *²*, 2805-2807.

⁽⁴⁾ Shimbo, K.; Tsuda, M.; Izui, N.; Kobayashi, J. *J. Org. Chem.* **²⁰⁰²**, *⁶⁷*, 1020-1023. (5) 1H NMR chemical shifts of the 6(9)-hemiacetal form (**1b**) were

assigned by the NOESY spectrum: δ_H 6.84 (H-3), 5.04 (H-11), 5.03 (H-16), 3.99 (H-15), 3.80 (H-9), 2.72 (H-4), 2.10 (H-8a), 1.96 (H-14a), 1.76 (H-5b), 1.60 (H-8b), 1.58 (H₃-25), 1.56 (H-5a), and 1.26 (H₃-23).

position	δc	$\delta_{\rm H}$	(m, Hz)
1	165.81, s		
$\boldsymbol{2}$	120.05, d	5.78	d. 15.6
3	153.56, d	6.59	dd, $9.5, 15.6$
$\overline{\mathbf{4}}$	32.07, d	3.06	m
$5a^a$	42.60, t	2.94	dd, 11.5, 17.8
$5b^a$		2.38	dd, $2.1, 17.8$
6	211.09, s		
$\overline{7}$	77.26 , s		
$8a^a$	44.94, t	1.97	d. 14.5
8b ^a		1.76	dd, $9.0, 14.5$
9	71.01, d	3.11	t, 9.0
10	39.23, d	2.25	m
11	128.61, d	4.86	m
12	138.21, s		
13	34.74, t	2.13^{b}	m
$14a^a$	33.97, t	1.86	m
$14b^a$		1.48	m
15	79.99, d	3.92	dt, $4.1, 11.0$
16	78.67. d	4.87	m
$17a^a$	42.67, t	2.10	dd, $7.4, 14.3$
$17b^a$		1.76	dd, $2.4, 14.3$
18	82.96, s		
19	44.85, t	1.47 ^b	m
20	17.82, t	1.32^{b}	m
21	14.55, t	0.91	t, 7.0
22	19.89, q	1.10 ^c	d, 6.6
23	26.58, q	1.35 ^c	S
24	16.84, q	0.87 ^c	d, 6.5
25	17.51, q	1.70 ^c	brs
26	25.74, q	1.23 ^c	${\bf S}$

TABLE 1. 1H and 13C NMR Data of 6-Keto Form (1a) of Amphidinolide Y (1) in CDCl3

^a a and b denote low-field and high-field resonances, respectively, of a geminal pair for C-5, C-8, C-14, and C-17. *^b* 2H. *^c* 3H.

FIGURE 1. Selected 2D (a) ${}^{1}H-{}^{1}H$ and ${}^{1}H-{}^{13}C$ and (b) ${}^{13}C-{}^{13}C$ correlations for amphidinolide X (1).

The geometry of the disubstituted olefin (C-2-C-3) in **a** was assigned as E by the $H^{-1}H$ coupling constant [*J*(H-2/H-3) 15.6 Hz]. Connections among partial structures **a** and **b** through a ketone carbonyl (C-6; δ_c 211.09) and an oxygenated quaternary carbon (C-7; δ_c 77.26) were assigned by HMBC correlations of H_2 -5 (δ_H 2.94 and 2.38)/C-6,H-8a (δ _H 1.97)/C-6, H-8a/C-7, H₃-23 (δ _H 1.35, 3H, s)/C-6, and H_3 -23/C-7. HMBC correlations for H_2 -13 (δ _H 2.13, 2H)/C-12 (δ _C 138.21), H₃-25 (δ _H 1.70)/C-12, and H-11 (δ _H 4.86)/C-13 (δ _C 34.74) suggested the presence of a trisubstituted double bond at C-11-C-12, which was assigned as *E*-geometry by NOESY cross-peaks for H-10 (*δ*^H 2.25)/H3-25 and H-11 (*δ*^H 4.86)/H2-13. Connectivities among C-17, C-19, and C-26 through C-18 were deduced from HMBC correlations of H-17a (δ_H 2.10)/C-18 (δ_C 82.96), H₂-19 (δ _H 1.47, 2H)/C-18, and H₃-26 (δ _H 1.23)/C-18. The existence of an ether linkage between C-15 and C-18 was implied by the NOESY correlation for H-15/

FIGURE 2. Rotation models for (a) C-9-C-10, (b) C-8-C-9, (c) C-7-C-8, and (d) C-4-C-5 bonds in the 6-keto form (**1a**) of amphidinolide Y (**1**).

 $H₃$ -26, thereby constructing a tetrahydrofuran ring with *syn*-relation for H-15/C-26 (Figure 2). The ester carbonyl (C-1) was shown to be adjacent to C-2 by an HMBC correlation of H-2 (δ _H 5.78) to C-1 (δ _C 165.81). The relatively lower-field resonance of H-16 (δ _H 4.87) suggested that C-16 was involved in the ester linkage with C-1. Although no HMBC correlation for H-16 to C-1 was observed, a two-bond correlation for C-16 (δ_c 78.67)/C-1 was observed in the 2D DEPT C-C LR Relay spectrum, 6 indicating that the ester linkage existed between C-1 and C-16. Thus, the gross structure of the 6-keto form of amphidinolide Y (**1**) was elucidated to be **1a**.

The relative stereochemistry of the tetrahydrofuran ring in **1a** was deduced to be H-15/H-16-anti and H-15/ C-26-syn, which were the same as those of the corresponding portion in **2**, from NOESY correlations as shown in Figure 2. The relative configurations at C-4, C-7, C-9, and C-10 of **1a** were elucidated on the basis of the *J*-based configuration analysis,⁷ and long-range ^{13}C -1H coupling constants obtained from the HETLOC8 and *J*IMPEACH- $MBC⁹$ spectra. The relative configuration for C-9–C-10 bond was assigned as erythro by *J*(H-9/H-10) (9.0 Hz), *^J*(C-11/H-9) (2.3 Hz), and *^J*(C-9/H-10) (-6.2 Hz) values as well as NOESY correlations for H -8a/ H_3 -24, H -8b/ H -10, H-9/H-11, and H-9/H₃-24 (Figure 2a).

(9) Williamson, R. T.; Marquez, B. L.; Gerwick, W. H.; Martin, G. E.; Krishnamurthy, V. V. *Magn. Reson. Chem.* **²⁰⁰¹**, *³⁹*, 127-132.

⁽⁶⁾ Fukushi, E.; Kawabata, J. Symposium Papers*.* In *43rd Symposium on The Chemistry of Natural Products*; Osaka, 2001, pp 341- 346.

⁽⁷⁾ Matsumori, N.; Kaneno, D.; Murata, M.; Nakamura, H.; Ta-

chibana, K. *J. Org. Chem.* **1999,** 64, 866–876.
(8) (a) Otting, G.; Wüthrich, K. *Quart. Rev. Biophys.* **1990**, *23*, 39–
96. (b) Wollborn, U.; Leibfritz, D. *J. Magn. Reson.* **1992**, *98*, 142–146.
(c) Kurz. M.: Schmiede (c) Kurz, M.; Schmieder, P.; Kessler, H. *Angew. Chem., Int. Ed. Engl.* **¹⁹⁹¹**, *³⁰*, 1329-1331.

FIGURE 3. NOESY correlations for the C-3-C-11 portion in the 6-keto form (**1a**) of amphidinolide Y (**1**).

FIGURE 4. Labeling patterns of the 6-keto form (**1a**) of amphidinolide Y (**1**) and amphidinolide X (**2**) resulting from feeding experiments with 13C-labeled acetates.

From analyses of rotation models for the C-8-C-9 and C-7-C-8 bonds (Figures 2b and 2c, respectively), the relative configuration of the 1,3-diol at C-7 and C-9 was assigned as a syn-relation (Figure 3). For the $C-4-C-5$ bond (Figure 2d), relations between C-6 and C-22 and between H-5b and C-3 were elucidated to be both anti. Since NOESY correlations were observed for H_2 -5/ H_3 -23 and H-5a/H-8a, conformation of the C-4-C-10 portion was elucidated as shown in Figure 4, in which the relative configurations at C-4, C-9, and C-10 of **1a** were the same as those of C-4, C-10, and C-11 in amphidinolide X (**2**).2

To determine the absolute stereochemistry of amphidinolide Y (1), application of modified Mosher's method¹⁰ for C -9 and oxidative cleavage of the C -6- C -7 bond were carried out as follows. $\Delta \delta$ values ($\Delta \delta = \delta_S - \delta_R$) obtained from 1H NMR data of 9-(*S*)- and 9-(*R*)-MTPA esters (**3a** and **3b**, respectively) showed negative values for H-4 (-0.02) , H₂-5 (both -0.02), H₂-8 (-0.02 and -0.11), and H_3-23 (-0.01), while those of H-10 (+0.07), H-11 (+0.06), and H_3 -24 (+0.14) were positive, suggesting that C-9 possessed *S*-configuration. Amphidinolide Y (**1**) was treated with lead tetraacetate to afford amphidinolide X (**2**), of which the spectral data including the CD spectrum $[\lambda_{\text{ext}} 224 \text{ nm } (\Delta \epsilon -7.9)]$ were the same as those of natural specimen of **2** [λ_{ext} 225 nm ($\Delta \epsilon$ -8.7)]. Therefore, the absolute configurations at seven chiral centers were assigned as 4*S*, 7*R*, 9*S*, 10*R*, 15*S*, 16*R*, and 18*R*.

Feeding experiments with [1-¹³C], [2-¹³C], and [1,2-¹³C₂] sodium acetates were carried out as reported previously,¹¹ and 13C-labeled amphidinolide Y (**1**) was separated by the same procedure as described before. The 13C NMR spectra (CDCl₃ and C_6D_6) of **1** derived from [1-¹³C] sodium acetate showed significant enrichment of 8 carbons (C-1, C-4, C-8, C-10, C-13, C-15, C-18, and C-20). On the other hand, enrichment by [2-13C] sodium acetate was observed for 18 carbons (C-2, C-3, C-5, C-6, C-7, C-9, C-11, C-12, C-14, C-16, C-17, C-19, C-21, C-22, C-23, C-24, C-25, and C-26), though intensity ratios for C-6, C-7, and C-12 were less than those of others. One-bond $C-C$ couplings for $C-1/$ C-2 (*J*_{CC}, 73 Hz), C-4/C-5 (*J*_{CC}; 32 Hz), C-8/C-9 (41 Hz), C-10/C-11 (44 Hz), C-13/C-14 (35 Hz), C-15/C-16 (40 Hz), C-18/C-19 (J_{CC} , 40 Hz), and C-20/C-21 (J_{CC} , 35 Hz) revealed the incorporation of eight intact acetates, suggesting the presence of two acetates from C-1 to C-2 and from C-4 to C-5 and three diketides from C-8 to C-11, from C-13 to C-16, and from C-18 to C-21. The C_1 branches at C-22, C-23, C-24, C-25, and C-26 were all derived from C-2 of acetates, in which the carbonyl carbons were lost. All 26 carbon signals contained in the 6-keto form (**1a**) of amphidinolide Y (**1**) were shown to be labeled by acetates (Figure 4). These incorporation patterns suggested that **1a** was generated from three diketide units, two acetate units, three isolated "**m**" units from C-2 of acetates, a "**m**-**m**" unit, and five branched C_1 units from C-2 of acetates. The labeling patterns at C-1-C-6 and C-23-C-7-C-21 parts in **1a** corresponded to those at diacid and diol units in amphidinolide X (**2**), indicating that **2** might be generated from **1a** through oxidative cleavage at the C-6-C-7 position of the minor 6(9)-hemiacetal form (**1b**).

The 6-keto form (**1a**) of amphidinolide Y (**1**) is a new 17-membered macrolide possessing a tetrahydrofuran ring, five branched methyls, a ketone, and two hydroxyl groups, and **1b** is 6(9)-hemiacetal isomer of **1a**. Amphidinolide Y (**1**) is considered to be a biogenetic precursor of amphidinolide X (**2**). Amphidinolide Y (**1**) exhibited cytotoxicity against murine lymphoma L1210 and human epidermoid carcinoma KB cells in vitro with IC_{50} values of 0.8 and 8.0 *µ*g/mL, respectively.

Experimental Section

Cultivation and Isolation. The dinoflagellate *Amphidinium* sp. (strain no. Y-42) was separated from the inside cells of the marine acoel flatworm *Amphiscolops* sp., which was collected off Sunabe, Okinawa. The dinoflagellate was unialgally cultured at 25 °C for 2 weeks in seawater medium enriched with 1% ES supplement and 13C-lableled sodium bicarbonate (100 mg/L). The harvested cells (315 g, wet weight, from 500 L of culture) were extracted with MeOH/toluene (3:1, 1 L \times 3). After addition of 1 M NaCl aq (1 L), the mixture was extracted with toluene (1 L \times 3). Parts (1.27 g) of the toluene-soluble fractions (3.78 g) were subjected to a silica gel column (CHCl₃/MeOH, 98:2) and a Sep-Pak cartridge C_{18} (MeOH/H₂O, 8:2) followed by C_{18} HPLC

⁽¹⁰⁾ Ohtani, I.; Kusumi, T.; Kashman, Y.; Kakisawa, H. *J. Am. Chem. Soc.* **¹⁹⁹¹**, *¹¹³*, 4092-4096.

⁽¹¹⁾ Tsuda, M.; Izui, N.; Sato, M.; Kobayashi, J. *Chem. Pharm. Bull.* **²⁰⁰²**, *⁵⁰*, 976-977.

[Mightysil RP-18, 5 *µ*m, Kanto Chemical Co., Inc., 10 × 250 mm; eluent, CH_3CN/H_2O (85:15); flow rate, 3 mL/min; UV detection at 220 nm] to afford amphidinolide Y (**1**, 2.3 mg, 0.0007%, wet weight, t_R 14 \sim 16 min) and together with amphidinolides G (0.0008%), H (0.0007%), W (0.009%), and X (**2**, 0.004%).

Amphidinolide Y (1): a colorless oil; $[\alpha]^{17}$ _D -33° (*c* 1.0, CHCl₃); UV (EtOH) λ_{max} 208 nm (*ε* 12 400); IR (neat) ν_{max} 3450, 2928, and 1711 cm⁻¹; FABMS m/z 433 (M - H₂O + H)⁺ and 451 (M ⁺ H)+; ESIMS *^m*/*^z* 451 (M + H)⁺ and 473 (M + Na)+; HRESIMS m/z 473.2875 [calcd for C₂₆H₄₂O₆Na (M + Na)⁺, 473.2879].

(*S***)-MTPA Ester (3a) of Amphidinolide Y (1).** To a CH2- Cl₂ solution (50 μ L) of amphidinolide Y (1, 0.18 mg) was added 4-(dimethylamino)pyridine (50 μ g), triethylamine (5 μ L), and (*R*)-(-)-MTPACl (1.5 *^µ*L) at 4 °C, and stirring was continued for 3 h. After addition of *N*,*N*-dimethyl-1,3-propanediamine (5 *µ*L) and evaporation of solvent, the residue was subjected to C_{18} HPLC (TSK-gel ODS-100S, Tosoh Co., Ltd., 4.6×250 mm; eluent CH₃-CN/H2O, 95:5; flow rate, 1.0 mL/min; UV detection at 230 nm) to give 9-(*S*)-MTPA ester (3a, 0.1 mg, t_R 12.6 min) of amphidinolide Y (1) as colorless oil: ¹H NMR (CDCl₃) δ 0.89 (3H, t, J = 7.0 Hz, H₃-21), 0.96 (3H, d, $J = 6.5$ Hz, H₃-24), 1.15 (3H, d, $J =$ 6.6 Hz, H3-22), 1.20 (3H, s, H3-26), 1.30 (2H, m, H2-20), 1.36 (3H, s, H₃-23), 1.40 (1H, m, H-8), 1.45-1.54 (3H, m, H-14 and H₂-19), 1.55 (1H, m, H-8), 1.66 (3H, s, H₃-25), 1.74 (1H, dd, $J = 2.0$, 14.2 Hz, H-17), 1.81 (1H, m, H-14), 2.12 (2H, m, H2-13), 2.14 $(1H, dd, J = 7.8, 14.2 Hz, H-17), 2.51 (1H, m, H-5), 2.55 (1H, m,$ H-5), 2.56 (1H, m, H-10), 3.00 (1H, m, H-4), 3.51 (3H, s, OCH3), 3.85 (1H, m, H-15), 4.95 (1H, brt, $J = 4.0$ Hz, H-16), 5.13 (1H, brd, $J = 8.0$ Hz, H-9), 5.50 (1H, d, $J = 8.5$ Hz, H-11), 5.72 (1H, d, $J = 15.6$ Hz, H-2), 6.52 (1H, dd, $J = 9.3$, 15.6 Hz, H-3), 7.39 (3H, m, Ph), and 7.62 (2H, m, Ph); FABMS *^m*/*^z* 667 (M + H)+; HRFABMS m/z 667.3435 [calcd for C₃₄H₅₀O₈F₃ (M + H)⁺, 667.3458].

(*R***)-MTPA Ester (3b) of Amphidinolide Y (1).** Amphidinolide Y (**1**, 0.44 mg) was treated with 4-(dimethylamino) pyridine (50 μ g), triethylamine (5 μ L), and (*S*)-(+)-MTPACl (1.5 μ L) by the same procedure as described above to afford the 9- (R) -MTPA ester (3b, 0.1 mg, t_R 12.4 min) of amphidinolide Y (1) as colorless oil: ¹H NMR (CDCl₃) δ 0.82 (3H, d, *J* = 6.5 Hz, H₃-24), 0.89 (3H, t, $J = 7.0$ Hz, H₃-21), 1.15 (3H, d, $J = 6.6$ Hz, H_3 -22), 1.23 (3H, s, H₃-26), 1.30 (2H, m, H₂-20), 1.37 (3H, s, H₃-23), 1.40-1.54 (4H, m, H-14, H-8, and H₂-19), 1.57 (1H, m, H-8), 1.61 (3H, s, H₃-25), 1.72 (1H, dd, $J = 2.0,14.2$ Hz, H-17), 1.76 $(1H, m, H-14)$, 2.06 $(2H, m, H₂-13)$, 2.12 $(1H, dd, J = 7.8, 14.2)$ Hz, H-17), 2.49 (1H, m, H-10), 2.53 (1H, m, H-5), 2.57 (1H, m, H-5), 3.02 (1H, m, H-4), 3.53 (3H, s, OCH3), 3.82 (1H, m, H-15), 4.94 (1H, brt, $J = 4.0$ Hz, H-16), 5.06 (1H, brd, $J = 8.0$ Hz, H-9), 5.44 (1H, d, $J = 8.5$ Hz, H-11), 5.72 (1H, d, $J = 15.6$ Hz, H-2), 5.44 (1H, d, $J = 8.5$ Hz, H-11), 5.72 (1H, d, $J = 15.6$ Hz, H-2), 6.53 (1H dd, $J = 9.3$, 15.6 Hz, H-3), 7.38 (3H, m, Ph), and 7.60 6.53 (1H, dd, *J* = 9.3, 15.6 Hz, H-3), 7.38 (3H, m, Ph), and 7.60
(2H m Ph): FABMS *m*/z667 (M + H)^{+,} HRFABMS *m*/z667 3467 (2H, m, Ph); FABMS *^m*/*^z* 667 (M ⁺ H)+; HRFABMS *^m*/*^z* 667.3467 [calcd for $C_{34}H_{50}O_8F_3$ (M + H)⁺, 667.3458].

Oxidation of Amphidinolide Y (1) with Lead Tetraacetate. Amphidinolide Y (**1**, 0.2 mg) was treated with lead tetraacetate (0.5 mg) in EtOAc (50 μ L) at room temperature for 15 h. After filtration of insoluble materials, the solvent was evaporated to afford amphidinolide X (**2**, 0.1 mg) as a colorless oil: CD (MeOH) $λ_{ext}$ 224 nm (Δε -7.9); FABMS *m*/*z* 449 (M + H)⁺; HRFABMS *m*/*z* 449.2918 [calcd for C₂₆H₄₁O₆ (M + H)⁺ 449.2903]. 1H NMR data of **2** were consistent with those of natural specimen.

Feeding Experiments with 13C-Labeled Precursors. The dinoflagellate cultured in a 100 L nutrient-enriched seawater medium was supplemented with $[1^{-13}C]$, $[2^{-13}C]$, or $[1,2^{-13}C_2]$ sodium acetate (610 μ M) in one portion at 4 days after inoculation, and then the culture was harvested by centrifugation after 14 days to obtain cells of the dinoflagellate (70 g as an average, wet weight). Extraction and isolation of amphidinolide Y (**1**) from the harvested cells were carried out through the same procedure as described above. The ¹³C-labeled amphidinolide \bar{Y} (1) was obtained in 0.001% yield as an average from wet weight of the cells.

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Supporting Information Available: General method, isotope incorporation results, spectral data of **1**, and an overlay of 1H NMR spectra of **2** derived from **1** and the natural specimen. This material is available free of charge via the Internet at http://pubs.acs.org.

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